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In solution chemistry, it is common to use H^+ as an abbreviation for the solvated hydrogen ion, regardless of the solvent. In aqueous solution H^+ denotes a solvated hydronium ion rather than a proton. The designation of an acid or base as "conjugate" depends on the context. The conjugate acid BH^+ of a base B dissociates according to $BH^+ + OH^- \rightleftharpoons B + H_2O$. which is the reverse of the e

[Alcalino - Wikipedia, la enciclopedia libre](#)

A fullerene is an allotrope of carbon whose molecule consists of carbon atoms connected by single and double bonds so as to form a closed or partially closed mesh, with fused rings of

five to seven atoms. The molecule may be a hollow sphere, ellipsoid, tube, or many other shapes and sizes. Graphene (isolated atomic layers of graphite), which is a flat mesh of regular hexagonal rings, can be ...

[??? - Wikipedia](#)

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